## **Communications**

## Lithium Niobate Microtubes within Ordered Macroporous Silicon by Templated Thermolysis of a Single Source Precursor

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Lithium niobate (LiNbO<sub>3</sub>) has been used for all-optical wavelength conversion<sup>1</sup> and ultrafast optical signal processing<sup>2-5</sup> because of its outstandingly rapid nonlinear optical response behavior, low switching power, and broad conversion bandwidth. Periodically poled photonic crystals<sup>6</sup> and inverse artificial opals made of LiNbO<sub>3</sub><sup>7</sup> have been reported. Ordered macroporous silicon (Si)8-10 is an extensively investigated two-dimensional photonic band gap material. Here we report on the preparation of hybrid systems consisting of LiNbO<sub>3</sub> and macroporous Si. Such composites are of considerable interest because of the integration of LiNbO<sub>3</sub> into silicon photonics. For instance, the band gap of macroporous Si could be tuned by poling LiNbO3 and thus changing its dielectric constant. Moreover, we investigated whether released LiNbO<sub>3</sub> tubes are accessible using macroporous Si as a sacrificial template. We have adapted a methodology involving the use of porous matrixes as molds, as investigated by Martin<sup>11-13</sup> and others.<sup>14-20</sup> We

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## Scheme 1. Molecular Structure of Lithium[Tetraethoxy(1-phenyl-1,3-butanedione)Niobate]

dropped a 5% solution of lithium [tetraethoxy(1-phenyl-1,3butanedione)niobate] (Scheme 1) in ethanol at ambient conditions on macroporous Si with a pore diameter,  $D_p$ , of 1  $\mu$ m. The synthesis and the properties of this single-source precursor for LiNbO<sub>3</sub> are described elsewhere.<sup>21</sup> The Si substrate into which the macropores were etched was a single-crystalline (100)-oriented wafer. We removed the residual liquid from the template surface. As the solvent evaporates, a layer of adsorbed molecules remains at the pore walls, which can be converted to the target compound by thermolysis. This procedure may be repeated once or several times. The conversion to crystalline LiNbO<sub>3</sub> was performed by annealing the wetted templates at 550, 650, and 750 °C, respectively, for 5 h. Then, the samples were cooled to room temperature at a rate of 2 K/min. Note that all these steps were performed under an argon atmosphere. The templates may be etched with 30 wt % aqueous KOH either partially at room temperature or completely at 70 °C.

Figure 1 represents LiNbO<sub>3</sub> tubes annealed at 550 °C, prepared by two successive wetting—crystallization—cooling cycles. Figure 1a shows a scanning electron microscopy (SEM) image of the openings of LiNbO<sub>3</sub> tubes partially embedded in the template. Their wall thickness is approximately 30 nm. A completely released LiNbO<sub>3</sub> tube is depicted in Figure 1b. Its aspect ratio (length/diameter) equals 100. The inset (top right of Figure 1b) shows its open end at higher magnification. Figure 1c is a bright field transmission electron microscopy (TEM) image, and Figure 1d is the corresponding dark field micrograph of a detail of a tube wall.

We performed X-ray diffraction (XRD) experiments in the reflection mode on aligned LiNbO<sub>3</sub> tubes located in the template pores. The samples were mounted in such a way that the long axes of the tubes were oriented parallel, and the template surface was perpendicular to the plane defined

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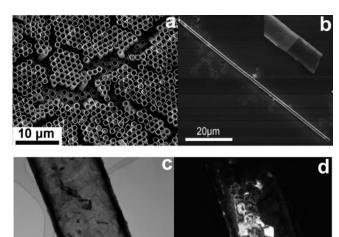
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200nm

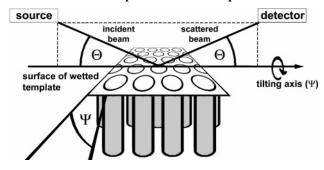


**Figure 1.** LiNbO<sub>3</sub> microtubes prepared by two successive wetting—crystallization—cooling cycles with an annealing temperature of 550 °C. SEM images of (a) openings of tubes attached to a partially removed template and (b) a released tube lying on a silicon substrate (inset: opening at higher magnification). (c) Bright-field and (d) dark-field TEM image of a tube wall.

200nm

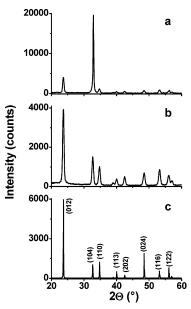
by the incident beam and the detector. The samples may also be tilted by an angle  $\Psi$  around an axis defined by the intersection of the template surface and the plane of the incident and the scattered X-ray beams. Both the  $\Psi$  and  $\Theta$  axes lie in the plane of the sample surface, enclosing a right angle (Scheme 2). All samples investigated showed the characteristic reflections of pure LiNbO3 independent of the selected thermolysis temperature and the number of wetting—crystallization—cooling cycles.

Scheme 2. Setup Used for XRD Experiments<sup>a</sup>



 $^a$  For the  $\Theta/2\Theta$  scans the samples were placed in the device in such a way that the surfaces of the templates were oriented perpendicularly to the plane defined by the incident and scattered X-ray beams. The samples may also be tilted by an angle  $\Psi$  around an axis defined by the intersection of the template surface and the plane of the incident and the scattered X-ray beams.

Figure 2a displays the  $\Theta/2\Theta$  scan of LiNbO<sub>3</sub> tubes within macroporous Si, fabricated by two consecutive wetting—crystallization—cooling cycles performed at 650 °C and one further cycle at 750 °C, with  $\Psi$  adjusted to 0°. For the  $\Theta/2\Theta$  scan of the same sample represented in Figure 2b, however, we selected a  $\Psi$  value of 10°. Figure 2c shows for comparison a powder diffractogram of LiNbO<sub>3</sub> measured on randomly oriented crystallites prepared by milling a single crystal of the ferroelectric noncentrosymmetric modification (space group R3c), fabricated by the Czochralski method,



**Figure 2.**  $\Theta/2\Theta$  scans on aligned LiNbO<sub>3</sub> tubes within macroporous Si, prepared by two wetting-crystallization-cooling cycles at 650 °C and one further wetting-crystallization-cooling cycle at 750 °C: (a)  $\Psi=0^{\circ}$  and (b)  $\Psi=10^{\circ}$ . (c) Powder pattern of the ferroelectric noncentrosymmetric modification of LiNbO<sub>3</sub>.

on a flat (100) Si wafer. The reflections were indexed according to JCPDS entry 78-0250. A conspicuous feature present in all diffractograms recorded with  $\Psi = 0^{\circ}$ , independent of the annealing temperature and the number of wetting-crystallization-cooling cycles, is that the strongest peak appears at  $2\Theta = 32.8^{\circ}$ . The corresponding XRD patterns of LiNbO<sub>3</sub>/macroporous Si hybrid materials prepared by one wetting-crystallization-cooling cycle, and two consecutive wetting-crystallization-cooling cycles, respectively, at 550 °C are shown in the Supporting Information (Figure S1). At  $2\Theta = 32.8^{\circ}$  the (104) reflection of LiNbO<sub>3</sub> and the (200) reflection of Si may coincide. The latter is a forbidden peak which only occurs when the Si lattice is distorted. Both XRD patterns displayed in Figure 2a and b show the characteristic reflections of pure LiNbO<sub>3</sub>. The striking difference is, however, that the very strong peak at 32.8° is not present in Figure 2b, which is similar to the powder pattern (Figure 2c), particularly regarding the relative peak intensities of the (012) and (104) reflections. Tilting the samples by  $10^{\circ}$  around the  $\Psi$  axis moves the scattered intensity originating from the {k00} reflections of macroporous Si out of the coverage of the detector. Therefore, we ascribe the strong reflection at  $2\Theta = 32.8^{\circ}$  occurring in the  $\Theta/2\Theta$ scans with  $\Psi$  adjusted to  $0^{\circ}$  predominantly to the (200) peak of Si. In the  $\Theta/2\Theta$  scan measured with  $\Psi$  adjusted to  $10^{\circ}$ the intensity in the corresponding  $2\Theta$  range exclusively originates from the (104) reflection of LiNbO<sub>3</sub>.

It is well-known that thermal oxidation of macroporous Si imposes considerable mechanical stress on Si forming the pore walls.<sup>22</sup> However, solely annealing macroporous silicon under argon does not result in the occurrence of lattice distortions abrogating the extinction of the (200) reflection (Supporting Information, Figure S2). Hence, the Si lattice

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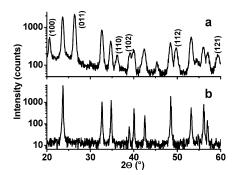


Figure 3.  $\Theta/2\Theta$  scan of aligned LiNbO<sub>3</sub> tubes within macroporous Si, prepared by two wetting-crystallization-cooling cycles at 650 °C under ambient conditions. The peaks that can be ascribed to quartz are denoted by the corresponding (hkl) values. (b) Powder pattern of the ferroelectric noncentrosymmetric modification of LiNbO<sub>3</sub>.

must have been distorted upon the conversion of the lithium [tetraethoxy(1-phenyl-1,3-butanedione)niobate] into LiNbO<sub>3</sub>.

Figure 3 shows the  $\Theta/2\Theta$  scan of LiNbO<sub>3</sub> tubes within macroporous silicon, recorded with  $\Psi = 10^{\circ}$ , after two wetting-crystallization-cooling cycles at 650 °C under ambient conditions. Besides the LiNbO<sub>3</sub> peaks, a second set of reflections occurs, which we ascribe to trigonal quartz (space group P3221, according to JCPDS entry 85-0794). This result indicates that the nature of the interface between LiNbO<sub>3</sub> and the pore wall is influenced by the presence or absence of oxygen.

Following the approach reported here a material with rapid nonlinear optical response behavior, low switching power,

and broad conversion bandwidth can be incorporated into a two-dimensional photonic crystal made of macroporous silicon. This may allow for switching the photonic band gap material by changing the polarization of LiNbO<sub>3</sub>. Along with the possibility to generate specific functional defect structures by omitting pores at specific positions, microscaled building blocks for high-speed optical switching and wavelength conversion may be accessible. The presented synthetic methodology is furthermore a generic approach for the fabrication of one-dimensional nano- and microstructures that can be adapted to any other functional material for which a suitable precursor system exists.

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**Supporting Information Available:**  $\theta/2\theta$  scans recorded with  $\Psi = 0^{\circ}$  on aligned LiNbO<sub>3</sub> tubes within macroporous Si prepared at 550 °C and powder pattern of the ferrolectric noncentrosymmetric modification of LiNbO<sub>3</sub> (Figure S1), and details of a  $\theta/2\theta$  scan on (100) oriented macroporous silicon annealed at 650 °C (Figure S2) (pdf). This material is available free of charge via the Internet at http://pubs.acs.org.

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